
Chemistry If8766 Ph And Poh Answers

ph and poh - newburyparkhighschool - ph and poh the ph of a solution indicates how acidic or basic that solution is. ph range of 0-7 acidic 7 neutral 7-14 basic since ... [h+] ph [oh-] poh acidic or basic 1. 10⁻⁵ m 5 10⁻⁹ m 9 acidic 2. 10⁻⁷m 7 10⁻⁷m 7 neutral 3. 10⁻¹⁰m 10 10⁻⁴ m 4 basic 4. 10⁻² m 2 10⁻¹²m 12 acidic 5. **ph and poh continued - newburyparkhighschool** - ph and poh continued 1. 0.01 m hcl [h+] = 0.01m ph = -log(0.01) = 2 2. 0.0010 m naoh [oh-] = 0.0010m poh = -log(0.010) = 3 → ph = 11 3. 0.050 m naoh

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